1) What is the density of 20.2 ml of a liquid which weighs 64.95 grams.
(Write the formula for Density. Circle and label the quantities given. Plug them into the formula. Solve for the unknown. Don't forget the unit!)
2) What is the mass of 10 ml of a liquid having a density of $2.15 \mathrm{~g} / \mathrm{ml}$ ?
3) What is the volume of 25 g of a solid that has a density of $15.37 \mathrm{~g} / \mathrm{ml}$ ?
4) How much does a $18.0 \mathrm{~cm}^{3}$ sample of pure silver weigh? Refer to Table $\mathbf{S}$ to find the density of silver, and then calculate its mass.
