

Dot's Crazy 2

Draw the Lewis (electron dot) structures of the following compounds.

REMEMBER: shared electrons are placed between the atoms involved.

1) SF_2 sulfur difluoride	4) CHCl_3 trichloromethane
2) PI_3 phosphorus trilodide	5) CSe_2 carbon diselenide
3) HF hydrogen fluoride	6) Br_2 bromine

Extras? O_3 , H_2O_2 , HCN , SO_2