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**Draw the Lewis (electron dot) structures** of the following compounds.

REMEMBER: shared electrons are placed between the atoms involved.

1) $\text{H}_2\text{S}$ hydrogen sulfide	4) $\text{CCl}_4$ carbon tetrachloride
2) $\text{PF}_3$ phosphorus trifluoride	5) $\text{CS}_2$ carbon disulfide
3) $\text{HF}$ hydrogen fluoride	6) $\text{O}_2$ oxygen